Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Deepak Chopra, A. R Choudhury and T. N. Guru Row*

Solid State and Structural Chemistry Unit, Indian Institute of Science, Bangalore 560 012, Karnataka, India

Correspondence e-mail: ssctng@sscu.iisc.ernet.in

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(\text{C-C}) = 0.002 \text{ Å}$ R factor = 0.041 wR factor = 0.109 Data-to-parameter ratio = 12.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3,4-Dimethoxyphenylacetic acid

The title compound, $C_{10}H_{12}O_4$, also known as homovaretic acid, forms hydrogen bonds, which generate chains rather than dimers in the crystal structure.

Received 14 February 2003 Accepted 3 March 2003 Online 14 March 2003

Comment

The title compound, (I), can be used as the starting material for the synthesis of a large number of 1,2,3,4-tetrahydroisoquinoline compounds (Nagarajan et al., 1985). The molecules (Fig. 1) form $O-H \cdots O$ hydrogen bonds, resulting in an infinite chain along the crystallographic b axis (Figs. 2 and 3). It is noteworthy that 3-methoxyphenylacetic acid crystallizes in the same space group, but forms $O-H \cdots O$ hydrogen-bonded dimers (Choudhury & Guru Row, 2002). The torsion angles C4-C3-C2-C1 [-100.1 (1)°], C3-C2-C1-01 [86.2 (2)°] and C3-C2-C1-O2 [-90.8 (2)°] differ from those in 3methoxyphenylacetic acid [88.1 (2), -0.2 (2) and 179.6 (1)°, respectively]. The two methoxy groups point away from each other [torsion angles $C9-O3-C5-C6 = 175.6 (1)^{\circ}$, C10- $O4-C6-C5 = 170.4 (1)^{\circ}$ and $O3-C5-C6-O4 = 1.3 (2)^{\circ}$]. The packing also involves two $C-H \cdots O$ hydrogen bonds (Table 1).



Experimental

The 98% pure compound was purchased from Sigma Aldrich. Single crystals were grown from a mixture of ethyl acetate and hexane at 283 K by slow evaporation.

Crystal data

$C_{10}H_{12}O_4$	$D_x = 1.305 \text{ Mg m}^{-3}$		
$M_r = 196.20$	Mo $K\alpha$ radiation		
Monoclinic, $P2_1/c$	Cell parameters from 541		
a = 14.258 (4) Å	reflections		
b = 7.185(2) Å	$\theta = 6.2-21.1^{\circ}$		
c = 9.773 (3) Å	$\mu = 0.10 \text{ mm}^{-1}$		
$\beta = 94.157 \ (5)^{\circ}$	T = 293 (2) K		
V = 998.5 (5) Å ³	Prism, colourless		
Z = 4	$0.60\times0.55\times0.55$ mm		
Data collection			
Bruker SMART CCD area-detector	2193 independent reflections		
diffractometer	1842 reflections with $I > 2\sigma(I)$		
φ and ω scans	$R_{\rm int} = 0.017$		
Absorption correction: multi-scan	$\theta_{\rm max} = 27.7^{\circ}$		
(SADABS; Bruker, 1998)	$h = -18 \rightarrow 17$		
$T_{\min} = 0.942, \ T_{\max} = 0.947$	$k = -9 \rightarrow 9$		
7814 measured reflections	$l = -12 \rightarrow 12$		

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 1

The molecular structure of the title compound, shown with 50% probability displacement ellipsoids.



Figure 2

Packing diagram of the title compound, viewed down the b axis.

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0519P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	+ 0.196P]
$wR(F^2) = 0.109$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} < 0.001$
2193 reflections	$\Delta \rho_{\rm max} = 0.22 \text{ e} \text{ \AA}^{-3}$
175 parameters	$\Delta \rho_{\rm min} = -0.19 \text{ e} \text{ \AA}^{-3}$
All H-atom parameters refined	

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O2-H20\cdots O1^i$	0.94 (3)	1.73 (3)	2.651 (2)	163 (3)
$C4-H4\cdots O1^{ii}$	0.940 (15)	2.583 (16)	3.473 (2)	158.2 (12)
$C8-H8\cdots O3^{iii}$	0.977 (15)	2.592 (15)	3.537 (2)	162.7 (13)
Symmetry codes: (i)	$1 - x, \frac{1}{2} + y, \frac{1}{2} - z$; (ii) $1 - x, y - \frac{1}{2}$,	$\frac{1}{2} - z$; (iii) $x, \frac{1}{2} - z$	$y, z - \frac{1}{2}$.



Figure 3 The hydrogen-bonded chains in (I), viewed down the *c* axis.

H atoms were located in a difference map and were refined isotropically. C–H distances are in the range 0.94 (2)–1.01 (2) Å and O–H is 0.95 (3) Å.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *PLATON* (Spek, 1998).

We thank the Department of Science and Technology, India, for data collection on the CCD facility set up under the IRHPA–DST programme.

References

- Altomare, A., Cascarano, G., Giacovazzo, C. & Guagliardi, A. (1993). J. Appl. Cryst. 26, 343–350.
- Bruker (1998). SMART, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.

Choudhury, A. R. & Guru Row, T. N. (2002). Acta Cryst. E58, 0889–0890. Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Nagarajan, K., Talwalker, P. K., Kulkarni, C. L., Shah, R. K., Shenoy, S. J. & Pravu, S. S. (1985). *Indian J. Chem. B*, **24**, 83–97.

Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.

Spek, A. L. (1998). PLATON. Utrecht University, The Netherlands.

Watkin, D. M., Pearce, L. & Prout, C. K. (1993). CAMERON. Chemical Crystallography Laboratory, University of Oxford, England.